

5. Solvent and Other Product Use

Greenhouse gas emissions are produced as a by-product of various solvent and other product uses. In the United States, emissions from Nitrous Oxide (N₂O) Product Usage, the only source of greenhouse gas emissions from this sector, accounted for less than 0.1 percent of total U.S. anthropogenic greenhouse gas emissions on a carbon equivalent basis in 2005 (see Table 5-1). Indirect greenhouse gas emissions also result from solvent and other product use, and are presented in Table 5-2 in teragrams of carbon dioxide equivalent (Tg CO₂ Eq.) and gigagrams (Gg).

Table 5-1: N₂O Emissions from Solvent and Other Product Use (Tg CO₂ Eq. and Gg)

Gas/Source	1990	1995	2000	2001	2002	2003	2004	2005
N ₂ O Product Usage								
Tg CO ₂ Eq.	4.3	4.5	4.8	4.8	4.3	4.3	4.3	4.3
Gg	14	14	15	15	14	14	14	14

Table 5-2: Indirect Greenhouse Gas Emissions from Solvent and Other Product Use (Gg)

Gas/Source	1990	1995	2000	2001	2002	2003	2004	2005
NMVOCs	5,216	5,609	4,384	4,547	3,911	3,916	3,921	3,926
CO	5	5	46	45	1	1	1	1
NO _x	1	3	3	3	5	5	5	5

5.1. Nitrous Oxide Product Usage (IPCC Source Category 3D)

N₂O is a clear, colorless, oxidizing liquefied gas, with a slightly sweet odor. N₂O is produced by thermally decomposing ammonium nitrate (NH₄NO₃), a chemical commonly used in fertilizers and explosives. The decomposition creates steam (H₂O) and N₂O through a low-pressure, low-temperature (500°F) reaction. Once the steam is removed through condensation, the remaining N₂O is purified, compressed, dried, and liquefied for storage and distribution. Two companies operate a total of five N₂O production facilities in the United States (CGA 2003).

N₂O is primarily used in carrier gases with oxygen to administer more potent inhalation anesthetics for general anesthesia and as an anesthetic in various dental and veterinary applications. As such, it is used to treat short-term pain, for sedation in minor elective surgeries and as an induction anesthetic. The second main use of N₂O is as a propellant in pressure and aerosol products, the largest application being pressure-packaged whipped cream. Small quantities of N₂O also are used in the following applications:

- Oxidizing agent and etchant used in semiconductor manufacturing;
- Oxidizing agent used, with acetylene, in atomic absorption spectrometry;
- Production of sodium azide, which is used to inflate airbags;
- Fuel oxidant in auto racing; and
- Oxidizing agent in blowtorches used by jewelers and others (Heydorn 1997).

Production of N₂O in 2005 was approximately 15 Gg. N₂O emissions were 4.3 Tg CO₂ Eq. (14 Gg) in 2005 (see Table 5-3). Production of N₂O stabilized during the 1990s because medical markets had found other substitutes for anesthetics, and more medical procedures were being performed on an outpatient basis using local anesthetics that do not require N₂O. The use of N₂O as a propellant for whipped cream has also stabilized due to the increased popularity of cream products packaged in reusable plastic tubs (Heydorn 1997).

Table 5-3: N₂O Emissions from N₂O Product Usage (Tg CO₂ Eq. and Gg)

Year	Tg CO ₂ Eq.	Gg
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1990	4.3	14
1995	4.5	14
2000	4.8	15
2001	4.8	15
2002	4.3	14
2003	4.3	14
2004	4.3	14
2005	4.3	14

Methodology

Emissions from N₂O product usage were calculated by first multiplying the total amount of N₂O produced in the United States by the share of the total quantity of N₂O attributed to each end use. This value was then multiplied by the associated emissions rate for each end use. After the emissions were calculated for each end use, they were added together to obtain a total estimate of N₂O product usage emissions. Emissions were determined using the following equation:

$$\text{N}_2\text{O Product Usage Emissions} = \sum_i [\text{Total U.S. Production of N}_2\text{O}] \times [\text{Share of Total Quantity of N}_2\text{O Usage by Sector } i] \times [\text{Emissions Rate for Sector } i]$$

where,

i = Sector.

The share of total quantity of N₂O usage by end use represents the share of national N₂O produced that is used by the specific subcategory (i.e., anesthesia, food processing, etc.). In 2005, the medical/dental industry used an estimated 89.5 percent of total N₂O produced, followed by food processing propellants at 6.5 percent. All other categories combined used the remainder of the N₂O produced. This subcategory breakdown has changed only slightly over the past decade. For instance, the small share of N₂O usage in the production of sodium azide has declined significantly during the decade of the 1990s. Due to the lack of information on the specific time period of the phase-out in this market subcategory, most of the N₂O usage for sodium azide production is assumed to have ceased after 1996, with the majority of its small share of the market assigned to the larger medical/dental consumption subcategory. The N₂O was allocated across these subcategories, a usage emissions rate was then applied for each sector to estimate the amount of N₂O emitted.

Only the medical/dental and food propellant subcategories were estimated to release emissions into the atmosphere, and therefore these subcategories were the only usage subcategories with emission rates. For the medical/dental subcategory, due to the poor solubility of N₂O in blood and other tissues, approximately 97.5 percent of the N₂O is not metabolized during anesthesia and quickly leaves the body in exhaled breath. Therefore, an emission factor of 97.5 percent was used for this subcategory (Tupman 2002). For N₂O used as a propellant in pressurized and aerosol food products, none of the N₂O is reacted during the process and all of the N₂O is emitted to the atmosphere, resulting in an emissions factor of 100 percent for this subcategory (Heydorn 1997). For the remaining subcategories, all of the N₂O is consumed/reacted during the process, and therefore the emissions rate was considered to be zero percent (Tupman 2002).

The 1990 through 1992 and 1996 N₂O production data were obtained from SRI Consulting's *Nitrous Oxide, North America* report (Heydorn 1997). These data were provided as a range. For example, in 1996, Heydorn (1997) estimates N₂O production to range between 13.6 and 18.1 thousand metric tons. Tupman (2003) provided a narrower range for 1996 that falls within the production bounds described by Heydorn (1997). These data are considered more industry specific and current. The midpoint of the narrower production range (15.9 to 18.1 thousand metric tons) was used to estimate N₂O emissions for years 1993 through 2001 (Tupman 2003). The 2002 and 2003 N₂O production data were obtained from the Compressed Gas Association *Nitrous Oxide Fact Sheet* and *Nitrous Oxide Abuse Hotline* (CGA 2002, 2003). These data were also provided as a range. For example, in 2003,

CGA (2003) estimates N₂O production to range between 13.6 and 15.9 thousand metric tons. Due to unavailable data, production for 2004 and 2005 were held at the value provided for 2003.

The 1996 share of the total quantity of N₂O used by each subcategory was obtained from SRI Consulting's *Nitrous Oxide, North America* report (Heydorn 1997). The 1990 through 1995 share of total quantity of N₂O used by each subcategory was kept the same as the 1996 number provided by SRI Consulting. The 1997 through 2001 share of total quantity of N₂O usage by sector was obtained from communication with a N₂O industry expert (Tupman 2002). The 2002 and 2003 share of total quantity of N₂O usage by sector was obtained from CGA (2002, 2003). Due to unavailable data, the share of total quantity of N₂O usage data for 2004 and 2005 was assumed to equal that of 2003. The emissions rate for the food processing propellant industry was obtained from SRI Consulting's *Nitrous Oxide, North America* report (Heydorn 1997), and confirmed by a N₂O industry expert (Tupman 2002). The emissions rate for all other subcategories was obtained from communication with a N₂O industry expert (Tupman 2002). The emissions rate for the medical/dental subcategory was substantiated by the *Encyclopedia of Chemical Technology* (Othmer 1990).

Table 5-4: N₂O Production (Gg)

Year	Gg
1990	16
1995	17
2000	17
2001	17
2002	15
2003	15
2004	15
2005	15

Uncertainty

The overall uncertainty associated with the 2005 N₂O emission estimate from N₂O product usage was calculated using the Intergovernmental Panel on Climate Change (IPCC) Good Practice Guidance Tier 2 methodology. Uncertainty associated with the parameters used to estimate N₂O emissions included that of production data, total market share of each end use, and the emission factors applied to each end use, respectively.

The results of this Tier 2 quantitative uncertainty analysis are summarized in Table 5-5. N₂O emissions from N₂O product usage were estimated to be between 4.1 and 4.5 Tg CO₂ Eq. at the 95 percent confidence level (or in 19 out of 20 Monte Carlo Stochastic Simulations). This indicates a range of approximately 4 percent below to 4 percent above the 2005 emissions estimate of 4.3 Tg CO₂ Eq.

Table 5-5: Tier 2 Quantitative Uncertainty Estimates for N₂O Emissions From N₂O Product Usage (Tg CO₂ Eq. and Percent)

Source	Gas	2005 Emission Estimate (Tg CO ₂ Eq.)	Uncertainty Range Relative to Emission Estimate ^a			
			(Tg CO ₂ Eq.)		(%)	
			Lower Bound	Upper Bound	Lower Bound	Upper Bound
N ₂ O Product Usage	N ₂ O	4.3	4.1	4.5	-4%	+4%

^a Range of emission estimates predicted by Monte Carlo Stochastic Simulation for a 95 percent confidence interval.

Recalculations Discussion

The N₂O production values for 2002, 2003, and 2004 have been updated relative to the previous Inventory based on revised production data presented in CGA (2003). The updated production data resulted in a decrease of 0.5 Tg

CO₂ Eq. (10 percent), respectively, in N₂O emissions from N₂O product usage for these years relative to the previous Inventory.

Planned Improvements

Planned improvements include a continued evaluation of alternative production statistics for cross verification and a reassessment of subcategory usage to accurately represent the latest trends in the product usage.

5.2. Indirect Greenhouse Gas Emissions from Solvent Use

The use of solvents and other chemical products can result in emissions of various ozone precursors (i.e., indirect greenhouse gases).¹ Non-CH₄ volatile organic compounds (NMVOCs), commonly referred to as “hydrocarbons,” are the primary gases emitted from most processes employing organic or petroleum based solvents. As some of industrial applications also employ thermal incineration as a control technology, combustion by-products, such as carbon monoxide (CO) and nitrogen oxides (NO_x), are also reported with this source category. In the United States, emissions from solvents are primarily the result of solvent evaporation, whereby the lighter hydrocarbon molecules in the solvents escape into the atmosphere. The evaporation process varies depending on different solvent uses and solvent types. The major categories of solvent uses include: degreasing, graphic arts, surface coating, other industrial uses of solvents (i.e., electronics, etc.), dry cleaning, and non-industrial uses (i.e., uses of paint thinner, etc.).

Total emissions of NO_x, NMVOCs, and CO from 1990 to 2005 are reported in Table 5-6.

Table 5-6: Emissions of NO_x, CO, and NMVOC from Solvent Use (Gg)

Activity	1990	1995	2000	2001	2002	2003	2004	2005
NO_x	1	3	3	3	5	5	5	5
Surface Coating	1	2	3	3	5	5	5	5
Degreasing	+	+	+	+	+	+	+	+
Graphic Arts	+	1	+	+	+	+	+	+
Dry Cleaning	+	+	+	+	+	+	+	+
Other Industrial Processes ^a	+	+	+	+	+	+	+	+
Non-Industrial Processes ^b	+	+	+	+	+	+	+	+
Other	NA	+	+	+	+	+	+	+
CO	5	5	46	45	1	1	1	1
Surface Coating	+	1	46	45	1	1	1	1
Degreasing	+	+	+	+	+	+	+	+
Graphic Arts	+	+	+	+	+	+	+	+
Dry Cleaning	+	1	+	+	+	+	+	+
Other Industrial Processes ^a	4	3	+	+	+	+	+	+
Non-Industrial Processes ^b	+	+	+	+	+	+	+	+
Other	NA	NA	+	+	+	+	+	+
NMVOCs	5,216	5,609	4,384	4,547	3,911	3,916	3,921	3,926
Surface Coating	2,289	2,432	1,767	1,863	1,602	1,604	1,606	1,608
Non-Industrial Processes ^b	1,724	1,858	1,676	1,707	1,468	1,470	1,472	1,474
Degreasing	675	716	316	331	285	285	286	286
Dry Cleaning	195	209	265	272	234	234	234	235
Graphic Arts	249	307	222	229	197	197	197	197

¹ Solvent usage in the United States also results in the emission of small amounts of hydrofluorocarbons (HFCs) and hydrofluoroethers (HFEs), which are included under Substitution of Ozone Depleting Substances in the Industrial Processes chapter.

Other Industrial Processes ^a	85		87		98	103	89	89	89	89
Other	+		+		40	42	36	36	36	37

^a Includes rubber and plastics manufacturing, and other miscellaneous applications.

^b Includes cutback asphalt, pesticide application adhesives, consumer solvents, and other miscellaneous applications.

Note: Totals may not sum due to independent rounding.

+ Does not exceed 0.5 Gg.

Methodology

Emissions were calculated by aggregating solvent use data based on information relating to solvent uses from different applications such as degreasing, graphic arts, etc. Emission factors for each consumption category were then applied to the data to estimate emissions. For example, emissions from surface coatings were mostly due to solvent evaporation as the coatings solidify. By applying the appropriate solvent-specific emission factors to the amount of solvents used for surface coatings, an estimate of emissions was obtained. Emissions of CO and NO_x result primarily from thermal and catalytic incineration of solvent-laden gas streams from painting booths, printing operations, and oven exhaust.

These emission estimates were obtained from preliminary data (EPA 2006), and disaggregated based on EPA (2003), which, in its final iteration, will be published on the National Emission Inventory (NEI) Air Pollutant Emission Trends web site. Emissions were calculated either for individual categories or for many categories combined, using basic activity data (e.g., the amount of solvent purchased) as an indicator of emissions. National activity data were collected for individual applications from various agencies.

Activity data were used in conjunction with emission factors, which together relate the quantity of emissions to the activity. Emission factors are generally available from the EPA's *Compilation of Air Pollutant Emission Factors, AP-42* (EPA 1997). The EPA currently derives the overall emission control efficiency of a source category from a variety of information sources, including published reports, the 1985 National Acid Precipitation and Assessment Program emissions inventory, and other EPA data bases.

Uncertainty

Uncertainties in these estimates are partly due to the accuracy of the emission factors used and the reliability of correlations between activity data and actual emissions.

